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87 Fr (223) 2 4 Be 9.01
12 Mg 24.31 20 Ca
40.08 38 Sr 87.62 56
Ba 137.34 88 Ra (226)
f/21 Sc 44.96 39 Y
88.91 57 f La 138.91
89 / Ac (227) 22 Ti
47.90 40 Zr 91.22 72
Hf 178.49 58 Ce
140.12 90 Th 232.04

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23 V 50.94 41 Nb

92.91 73 Ta 180.95 59

Pr 140.91 91 Pa 231.04

Atomic Number 24 Cr

52.00 42 Mo 95.94 74

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47.90 40 Zr 91.22 72
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2.4 10^{-9} > 1.4×10^{-16}
AND precipitate will form If value given is used: ALTERNATIVE 3:

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«Q [Pb²⁺] [OH⁻]² 2.4
10⁻⁴ (1.0 10⁻²)²» 2.4
10⁻⁸ Q > K_{sp} AND
precipitate will form OR
2.4 10⁻⁸ > 1.43 10⁻²⁰
AND precipitate will
form 2

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